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energy to support your answer answer
endothermic the products have more
energy than the reactants b on the
diagram provided in your answer
booklet draw a dashed line to indicate a
potential energy curve for the reaction
if a catalyst is added 46 given the
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potential energy than the products have
and the extra energy is released to the
surroundings it is called activation
energy the height of the hill on a
potential energy diagram is a measure
of the endothermic the reaction above
can be categorized as

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web this is like a one dimensional
system whose mechanical energy e is a
constant and whose potential energy
with respect to zero energy at zero
displacement from the spring s
unstretched length x_0 is $u_x = \frac{1}{2} kx^2$
figure 8 5 2 8 5 2 a glider between
springs on an air track is an example of
a horizontal mass spring system

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and the pe of the products is 190kj
what is the Δh 2 if the pe of the
reactants is 430kj and the activation
energy is 220 how much total energy
will be needed to be added for the
reaction to proceed transcribed image

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